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(54) Title: COMPOSITION FOR TREATING HAIR			
(57) Abstract			
<p>The present invention relates to a hair treatment composition which provides an excellent finishing effect and superior protection against environmental, chemical, and grooming-associated damage. The compositions of the present invention comprise hydrolyzed protein having an abundance of anionic amino acids and in particular, sulphur-containing amino acids, as well as divalent cationic compounds, such that the anionic components of the hydrolyzed protein may effectively bind to the hair by means of cationic bridges. While bound to the hair, the sulphur containing amino acids in the hydrolyzed protein may serve as "decoys" for the effects of a variety of damaging agents. Compositions of the present invention may further comprise a vitamin compound which enhances these protective benefits.</p>			
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- 1 -

Description

COMPOSITION FOR TREATING HAIR

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I. INTRODUCTION

The present invention relates to a hair treatment composition which provides an excellent finishing effect and superior protection against environmental, chemical and grooming-associated damage. The compositions of the present invention comprise hydrolyzed protein having an abundance of anionic amino acids and in particular, sulphur-containing amino acids, as well as divalent cationic compounds, such that the anionic components of the hydrolyzed protein may effectively bind to the hair by means of cationic bridges. While bound to the hair, the sulphur containing amino acids in the hydrolyzed protein may serve as "decoys" for the effects of a variety of damaging agents. The compositions of the present invention may further comprise a vitamin compound which enhances these protective benefits.

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BACKGROUND OF THE INVENTION

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Human hair is comprised of keratin ("hair keratin"), a tough fibrous protein having a high sulfur content. Hair keratin is routinely exposed to a variety of activities that compromise its structure. For example, simple grooming and styling results in loss of cuticle scale, subjecting the hair's inner cortex to further damage and eventually leading to hair breakage. Shampooing causes frictional damage to cuticle

- 2 -

scale and extracts hair protein. Ultraviolet light from the sun reduces the hair's mechanical strength and causes fading of natural and applied hair color. Exposure to chlorinated water in swimming pools causes oxidation of hair keratin, leading to increased protein loss during combing and brushing.

5 These practices are less damaging to hair than direct exposure to the reactive chemicals commonly used by hair stylists, such as permanent wave agents, hair colors, bleaches, straighteners, peroxides, thio solutions, sodium hydroxide and the like. However, the cumulative effects of daily cleansing, grooming, and ultraviolet light are far greater than the damage associated with the above-mentioned chemicals. However, if
10 hair has been chemically treated, then daily grooming, shampooing and sun exposure has an even greater potential for causing extensive damage.

 In view of the above, there is a need for a novel class of hair treatment compositions which provide not only beautifying effects to damaged and non-damaged hair, but also provide superior protection from the various activities mentioned above that
15 compromise hair structure.

SUMMARY OF THE INVENTION

 The present invention relates to a composition comprising an effective
20 amount of an anionic hydrolyzed protein containing an abundance of sulphur-containing amino acids, which effectively binds to the anionic surface of hair via divalent cationic bridges. Such a composition, when applied to the hair, reduces hair damage and protein loss caused by grooming, excessive heat, chlorinated water, ultraviolet light from the sun, and reactive chemicals. The addition of a vitamin compound which has antioxidant
25 and/or ultraviolet absorbent properties further enhances the protective efficacy of the compositions of the present invention.

 In particular, the present invention provides for a hair treatment composition comprising:

- 3 -

(a) from about 0.1 to about 50% by weight of a hydrolyzed protein, wherein the molar ratio of anionic amino acids to cationic amino acids is at least about 1.1 to 1.0, and wherein the hydrolyzed protein has an average molecular from about 200 to 500,000 daltons and comprises at least 0.25% by weight of sulphur-containing amino acids;

(b) a divalent cationic compound, wherein the molar ratio of said compound to the anionic amino acids of the hydrolyzed protein is from about 0.2:1 to about 4:1;

(c) optionally from about 0.01 to about 5% by weight of a vitamin compound, wherein the vitamin compound absorbs ultraviolet light within the wavelength region between 290 nm and 420 nm and/or is effective in retarding oxidation of hair; and

(d) from about 1 to 99.89% by weight of a cosmetic carrier.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1a is a bar graph illustrating the change in lightness of irradiated hair swatches treated with solutions containing varying molar ratios of divalent cationic compounds to hydrolyzed protein from wheat.

Figure 1b is a bar graph illustrating the change in color of irradiated hair swatches treated with solutions containing varying molar ratios of divalent cationic compounds to hydrolyzed protein from wheat.

Figure 1c is a bar graph illustrating the change in lightness of irradiated hair swatches treated with varying molar ratios of divalent cationic compounds to hydrolyzed protein from milk.

Figure 1d is a bar graph illustrating the change in color of irradiated hair swatches treated with varying molar ratios of divalent cationic compounds to hydrolyzed protein from milk.

- 4 -

Figure 2 is a bar graph illustrating the mechanical strength of irradiated hair treated with hydrolyzed protein alone or in combination with divalent cationic compounds.

5 Figure 3 is a bar graph illustrating the effect of chlorinated water on hair treated with hydrolyzed protein alone or in combination with divalent cationic compounds.

Figure 4 is a bar graph illustrating the effect of extended periods of irradiation to hair treated with either hydrolyzed protein alone or in combination with divalent cationic compounds.

10 Figure 5 is a bar graph comparing the protective effects of hydrolyzed protein from wheat and hydrolyzed protein from collagen, which were either succinylated or sulfonated.

Figure 6 is a bar graph illustrating the effect of irradiating hair treated with hydrolyzed protein, divalent cationic compounds and vitamin compounds.

15 Figure 7 is a bar graph illustrating the efficacy of hydrolyzed protein, divalent cationic compounds and vitamin compounds in retarding protein loss from hair.

Figure 8 is a three dimensional graph illustrating the mechanical strength of irradiated hair treated with hydrolyzed protein, divalent cationic compounds and vitamin compounds.

20 Figure 9 is a three dimensional graph illustrating the mechanical strength of irradiated and combed normal, bleached and permed hair treated with hydrolyzed protein, divalent cationic compounds and vitamin compounds.

25 Figure 10 is a bar graph illustrating the change in color of irradiated hair swatches treated with solution containing varying molar ratios of dibasic cationic amino acid compounds to hydrogen protein.

- 5 -

DETAILED DESCRIPTION OF THE INVENTION

The present invention relates to compositions for treating hair which protect hair from photo damage, chemical damage and protein loss due to grooming activities including, but not limited to, combing, brushing, blow drying, curling and the like. Such compositions of the invention comprise a hydrolyzed protein, a divalent cationic compound, and preferably a vitamin compound, in a cosmetic carrier.

The compositions according to the present invention include, but are not limited to, shampoos, conditioners, hair treatment creams, styling gels, mousse, pump hair sprays and aerosol hair sprays, set lotions, blow styling lotions, hair color lotions, hair relaxing compositions, permanent wave first agents, and permanent wave second agents.

The amount of hydrolyzed protein in these compositions is dependent on various considerations, such as product esthetics and the type of commercial product end-use desired, for example, as a shampoo, cream or treatment conditioner for normal hair, oily hair, dry hair, permed hair, thick hair, fine hair, etc. The considerations regarding the appropriate amount of hydrolyzed protein to be used in view of the end-use would be known to those of ordinary skill in the art. In this regard, products such as shampoos utilize relatively less hydrolyzed protein, generally in the range of 0.1 - 1.0% by weight of the composition, as the substantivity from shampoos is lower due to their detergency. High concentrations of protein in shampoos tend to adversely affect foam characteristics, and excess protein is merely washed away during rinsing. However, treatment products designed to build strength into the hair have longer hair contact times and high concentrations of protein are utilized to favor absorption. Such compositions may contain up to 50% by weight of hydrolyzed protein.

Thus, compositions of the present invention comprise from about 0.1 to 50.0% by weight, preferably from about 0.1 to 10.0% by weight, and more preferably from about 0.1 to 4.0% by weight of a hydrolyzed protein.

- 6 -

The term "hydrolyzed protein," as used herein, refers to the product of the hydrolysis of homogeneous or heterogeneous proteins, or their respective components, derivatives or combinations thereof, from sources including, but not limited to, plants and their respective components, seeds, animal bones, connective tissue, animal keratin, bovine and porcine collagen, human hair, wool, silk, elastin, reticulin, milk, egg, wheat, corn, soya, oats, casein, albumin, or any collagenous or keratinic substance, or derivatives thereof. With respect to the compositions of the present invention, there is no restriction on the source of protein, protein components or derivatives, provided that the hydrolyzate has the desired physical properties described herein. However, preferred protein sources include keratin, soya, milk, collagen, wheat and their respective components, derivatives or combinations thereof. Additional preferred protein sources may be identified as proteins having desirable amino acid compositions, as set forth below. Suitable derivatives include proteins chemically modified to alter their ionic charge, for example, by sulfonation or succinylation.

The compositions of the present invention are not limited to hydrolyzed proteins produced from naturally occurring proteins. Synthetic proteins, peptides, or amino acids as well as naturally occurring proteins, peptides or amino acids or mixtures of naturally occurring and synthetic proteins and/or peptides and/or amino acids may also be used according to the invention. Hydrolyzed protein prepared from various proteins, their respective components, and derivatives may be combined and used in the composition of the present invention. Moreover, a hydrolyzed protein may be supplemented by the addition of one or more natural or synthetic peptides or amino acids to achieve the ionic characteristics or sulphur content discussed below.

Methods for producing hydrolyzed proteins from the above-mentioned protein sources include, but are not limited to: 1) acid hydrolysis; 2) alkali hydrolysis; and 3) enzyme hydrolysis using a suitable protease. These methods, along with several others, for preparing hydrolyzed proteins are well known in the art. Further, hydrolyzed proteins suitable for the compositions of the present invention are commercially available.

- 7 -

Hydrolyzed proteins typically have an average molecular weight from about two hundred to several hundred thousand daltons, depending on the nature of the protein and/or the extent of hydrolysis. While hydrolyzed proteins of low molecular weight favor absorption, they are more easily lost from the hair by subsequent washing. On the other hand, proteins of very high molecular weight do not penetrate well. Generally, an average molecular weight of 500,000 daltons or less is desirable for hydrolyzed proteins utilized in this invention.

Therefore, in particular embodiments of the present invention, the average molecular weight of the hydrolyzed protein is from about 200 to 500,000 daltons, preferably from about 200 to 100,000 daltons, more preferably from about 200 to 50,000 daltons, and even more preferably from about 200 to 20,000 daltons.

In still further embodiments of the invention, the above-mentioned proteins or their derivatives, synthetic peptides, natural peptides or combinations thereof, may be comprised in the composition of the present invention without any additional hydrolysis treatment, provided that the desired molecular weight, ionic characteristics and sulphur content set forth below may be achieved.

Every protein molecule can be considered as a polymer of amino acids. There are 20 naturally occurring varieties of amino acid, each with a common backbone combined with one of 20 variable side chains. Amino acids are commonly classified as either neutral, anionic or cationic amino acids based on the charge of the variable side chain.

The naturally occurring anionic amino acids include aspartic acid and glutamic acid, both of which have a carboxylate group at the termini of their respective variable side chains. The present invention also provides for amino acids modified such that the net charge of the amino acid is made anionic, or if the net charge is already anionic, such charge may be increased. Examples of such modifications include, but are not limited to, succinylation or sulfonation of the amino acid. All such amino acids are considered, herein, to be "anionic amino acids."

- 8 -

The naturally occurring cationic amino acids include lysine, arginine and histidine, all of which comprise an amino group in their respective variable side chains. Amino acids may also be modified to result in a net cationic charge. All such amino acids are considered herein to be "cationic amino acids".

5 Hair care products are generally in the pH range of 2.0 to 10.0. Proteins having a predominance of anionic amino acids, relative to cationic amino acids, are likely to be negatively charged at the general pH range of hair care products. Conversely, proteins having a relative abundance of cationic amino acids are likely to be positively charged under similar conditions.

10 It is preferred that the amino acid composition of the hydrolyzed protein in the present invention contain a predominance of anionic amino acids relative to cationic amino acids, so that the hydrolyzed protein at the general pH range of hair care products is negatively charged.

 Specifically, the composition of the present invention preferably
15 comprises a hydrolyzed protein, wherein the molar ratio of anionic amino acids to cationic amino acids is at least 1.1:1, preferably from about 1.1:1 to 1.1:1 to 25:1, and more preferably from about 1.1 to 1.1:1 to 15:1.

 The term "molar ratio of anionic amino acids to cationic amino acids" of
20 the hydrolyzed protein in the present invention means the total molar amount of anionic amino acids (for example, adding the individual molar amount of the anionic amino acids, such as aspartic acid and glutamic acid) in relation to the total molar amount of cationic amino acids (for example, adding the individual molar amount of lysine, arginine and histidine) in the hydrolyzed protein.

 The molar ratio of anionic amino acids to cationic amino acids may be
25 readily calculated based on the overall amino acid composition of the hydrolyzed protein or mixtures of hydrolyzed proteins. Such compositions are frequently known in the art. Alternatively, the amino acid composition of any protein may be determined using techniques known to those skilled in the art, which include, but are not limited to, (1) automated amino acid analysis and (2) high pressure liquid chromatography. Such

- 9 -

techniques separate the individual amino acids, and the molar ratio of anionic to cationic amino acids may be determined.

Further, the hydrolyzed protein of the present invention is characterized in that it comprises at least 0.25%, preferably from about 0.25 to 15%, and more preferably from about 0.25 to 5.0% by weight of a sulphur containing amino acid. The "term sulphur-containing amino acid" as used herein refers to any amino acid, natural or synthetic, containing sulphur in any form, including, but not limited to, sulphydryl groups or disulphide bonds. Sulphur-containing amino acids suitable for the present invention include, but are not limited to, cysteine, cystine, methionine, and their respective derivatives and synthetic analogues. Compositions according to the present invention may, in particular embodiments, comprise in addition to the protein hydrolyzate, sulphur-containing compounds such as amino acids or sulphur-containing salts which may supplement the amount of sulphur in the composition to achieve the percentages (by weight, relative to the hydrolyzed protein) set for the above.

Comparative tests that have been carried out, and are described below, suggest that the sulphur-containing amino acids in the hydrolyzed protein shield the hair from various assaults, such as exposure to ultraviolet light or chemicals known to cause hair damage. Without being bound by any theory, it is believed that the sulphur-containing amino acids act as a buffer or "decoy" between the hair and the assaulting agent.

For example, sunlight and certain chemical agents damage hair by breaking naturally occurring disulphide bonds. Applying proteins comprising sulphur-containing amino acids to hair provides more sulphur bonds, particularly disulphide bonds, for breakage by exposure to sunlight or chemical treatments, thus leaving the sulphur bonds occurring naturally in the hair less vulnerable to such assaults.

In that regard, additional compounds having sulphydryl groups or disulphide bonds may be added to the compositions of the present invention to enhance protection from oxidation and ultra-violet light. For example, free residual cystine, such as cystine diester hydrochloride or any other sulphur-containing salt, may be added to

- 10 -

increase the content of disulphide bonds present in the composition. Alternatively, peptide fragments rich in sulphur-containing amino acids or any other compound having sulphhydryl groups or disulphide bonds may be used in the present invention.

Synthetic or naturally occurring proteins and hydrolyzed proteins suitable for the present invention can be readily prepared from appropriate sources, or, alternatively, purchased commercially.

Table 1 sets forth relevant characteristics of nonlimiting examples of preferred protein sources from which the hydrolyzed proteins of the present invention may be prepared. These proteins were purchased from Croda, Inc. of Edison, New Jersey.

TABLE 1

Brand Name	Protein Source	MW	% Cystine	Molar Ratio (anionic/ cationic amino acids)
Crotein WKP	Hydrolyzed protein from keratin	600	10.22	1.1:1
Hydrosoy 2000/SF	Hydrolyzed protein from soy	1,000	1	2.2:1
Hydrolactin 2500	Hydrolyzed protein from milk	2,500	0.4*	2.19:1
Crotein SPA,O,C	Hydrolyzed protein from collagen	10,000	0.5	1.6:1
Hydrotriticum 2000	Hydrolyzed protein from wheat	3,000	1.80	6.2:1
Cropeptide W	Hydrolyzed protein from wheat	2,500	1.40	13.9:1
			*Cysteine	

Hair care products are generally in the pH range of 2.0 to 10.0, although products for shampooing, conditioning and treatment of the hair are more generally in the range of 3.5 to 10.0, and even more typically between the pH of about 3.5 and 8.0. At a

- 11 -

pH greater than 3.8, the surface of the hair fiber has a predominance of negative charges, so that the divalent cationic compound is more effective at assisting with protein binding. In this regard, the preferred pH of the composition of the present invention is between 2.0 to 10.0, preferably from about 3.5 to about 10.0, and more preferably from about 3.5 to about 8.0.

The compositions of the invention also comprise a divalent cationic compound, for example, a divalent mineral cation, including, but not limited to, divalent calcium, copper, magnesium, manganese, iron, strontium, zinc, cadmium, barium, silver, nickel, cobalt or mercury. The divalent mineral may preferably be obtained from an ionic salt, including, but not limited to, salts formed with carboxylate, sulfonate, phosphate, halide, carbonate, silicate, nitrate, or pantothenate ions.

Further, divalent non-mineral compounds may be used in the composition of the present invention, including, but not limited to, a peptide that comprises at least two cationic amino acids, such as, for example, histidine, arginine, lysine, ornithine hydroxylysine, their respective derivatives, analogues or combinations thereof, such that the peptide has a net divalent cationic charge. The divalent cationic charge may result from the presence of cationic amino acids of a single type, or multiple types. Such a peptide may comprise from at least 2 to about 200 amino acids, preferably between 2 to 10 amino acids. The molar ratio of the divalent non-mineral compound to anionic amino acids of the hydrolyzed protein is at least 0.2:1 to 4:1, preferably between about 0.5:1 to 1:1, and more preferably about 0.5:1. Stated another way, the ratio of the cationic amino acids of the divalent non-mineral compound to the anionic amino acids of the hydrolyzed protein is at least 0.4:1 to 8:1, preferably between about 1:1 to 2:1, and more preferably at about 1:1.

Additionally, peptides having a net cationic charge greater than 2 may be used in the composition of the present invention, including, but not limited to, a peptide that comprises at least two cationic amino acids, such as, for example, histidine, arginine, lysine, ornithine hydroxylysine, their respective derivatives, analogues or combinations

- 12 -

thereof. Such a peptide may comprise from at least 2 to about 200 amino acids, preferably between 2 to 10 amino acids. The ratio of the cationic amino acids of the peptide to the anionic amino acids of the hydrolyzed protein is at least 0.4:1 to 8:1, preferably between about 1:1 to 2:1, and more preferably at about 1.0:1.

5 Other divalent non-mineral substances which may be used in the compositions of the present invention, include, but not are not limited to, diamino compounds having the general formula:



Wherein n=1-6 and R₁, R₂, R₃, R₄, and R₅, which may be the same or different, may be a hydrogen or an alkyl group, including but not limited to, a methyl, ethyl, isopropyl, butyl, or propyl group, such that the compound has a net divalent cationic charge. The molar ratio of the diamino compound to the anionic amino acids of the hydrolyzed protein is at least 0.2:1 to 4:1, preferably between about 0.5:1 to 1:1, and more preferably at about 1:1. Stated another way, the ratio of the amino groups of the diamino compound to the anionic amino acids of the hydrolyzed protein is at least 0.4:1 to 8:1, preferably between 1.0:1 to 2:1, and more preferably at about 1:1.

20 Other cationic polyamino compounds which may be used in the compositions of the present invention include compounds having at least three amino groups. Such cationic polyamino compounds may include compounds having the general formula:



30 Wherein n=1-6 and R₁, R₂, R₃, R₄, and R₅, which may be the same or different, may be a compound comprising one or more amino groups or a hydrogen or an alkyl group, including but not limited to, a methyl, ethyl, isopropyl, butyl, or propyl group. The ratio of the amino groups of the cationic polyamino compound to the anionic amino acids of the hydrolyzed protein is between about 0.4:1 to 8:1, preferably between about 1.0:1 to

- 13 -

2:1, and more preferably about 1:1. Examples of such compounds which may be used in the present invention, include, but are not limited to, polyethylenimine, $(C_2H_5N)_n$, where n = from about 1-100, or greater. One brand-name of this compound is POLYMINEPEI®, sold by BASF of New Jersey. Its molecular weight is between about 800 to 750,000. Another compound comprising at least three amino groups and which may be used in the present invention includes polyamidoamine. Such compound is sold under the brand-name STARBURST PAMAM by Dendritech, Inc. of Michigan.

Other polyamino cationic compounds which may be used in the compositions of the present invention include cationic amino acid compounds. The term “cationic amino acid compound,” as used in the specification and claims, is defined to include within the scope of that term any amino acid having an amino group in its variable side chain, such as, for example, histidine, lysine, arginine, hydroxylysine, ornithine, their respective derivatives, analogues, and stereoisomeric configurations or combinations thereof. Derivatives of the cationic amino acid compound may include salt forms of the amino acid. For example, the cationic amino acid of the present invention may comprise a salt selected from the group consisting of hydrochloride, hydrochloride monohydrate, monohydrochloride, monohydrate, methyl ester dihydrochloride, dihydrochloride, ethyl ester dihydrochloride or combinations thereof.

Specific cationic amino acid compounds which may be suitable for the present invention include the following:

Arginine

Arginine aspartate	DL-Arginine hydrochloride monohydrate
Arginine PCA	D-Arginine monohydrochloride
L-Arginine	L-Arginine monohydrochloride
DL-Arginine	L-Homoarginine hydrochloride
D-Arginine hydrate	
D-Arginine hydrochloride	
L-Arginine hydrochloride	
L-Arginine methyl ester dihydrochloride	

- 14 -

Histidine

5	D-Histidine	L-Histidine methyl ester dihydrochloride
	L-Histidine	D-Histidine monohydrochloride monohydrate
	DL-Histidine	L-Histidine monohydrochloride monohydrate
		DL-Histidine monohydrochloride monohydrate

Lysine

10	Lysine	L-Lysine dihydrochloride
	DL-Lysine	L-Lysine monohydrate
	L-Lysine	DL-Lysine dihydrochloride
	D-Lysine	L-Lysine ethyl ester dihydrochloride
15	Lysine aspartate	
	Lysine PCA	
	L-Lysine, compound, with 5-oxo-L-proline (1:1)	
	D-Lysine hydrate	
20	L-Lysine hydrate	
	DL-Lysine monohydrochloride	
	D-Lysine monohydrochloride	
	L-Lysine monohydrochloride	

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Hydroxylysine

30	5-Hydroxy-DL-lysine hydrochloride
	5-Hydroxy-L-lysine dihydrochloride monohydrate
	5-Hydroxylysine hydrochloride

Ornithine

35	D-Ornithine hydrochloride
	L-Ornithine hydrochloride
	L-Ornithine dihydrochloride
	D-Ornithine monohydrochloride
	L-Ornithine monohydrochloride

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The molar ratio of the cationic amino acid compound to anionic amino acids of the hydrolyzed protein is at least 0.2:1 to 4:1, preferably between about 0.5:1 to

- 15 -

1:1, more preferably at about 0.5:1. Stated another way, the ratio of the amino groups of the cationic amino acid compound to anionic amino acids of the hydrolyzed protein is at least 0.4:1 to 8:1, preferably between about 1.0 and 2:1, and more preferably at about 1:1.

5 It is desirable that the molar ratio of divalent compound to the anionic amino acids of the hydrolyzed protein is effective in "bridging" the hydrolyzed protein to the hair to form a hydrolyzed protein layer, wherein said layer is effective in at least partially protecting the hair from photo damage, oxidation and grooming-associated damage. In this regard, the preferred molar ratio of divalent cationic compound to anionic amino acids of the hydrolyzed protein is between about 0.2:1 to 4:1, preferably
10 between about 0.5:1 to 1:1, and more preferably at about 0.5:1.

Examples 1-5, below, demonstrate the correlation between the molar ratio of divalent cationic compound to the anionic amino acids of the hydrolyzed protein and the protective effects of the compositions of the present invention.

The composition of the invention may further comprise from about 0.01 to
15 5% by weight of a vitamin compound or its respective derivative which functions as an antioxidant and/or absorbs ultraviolet light. Preferably, the antioxidant vitamin is more readily oxidized than hair and hence can retard or inhibit the oxidation of the hair when applied. The vitamin compound may alternatively or additionally absorb ultra-violet light within the wavelength region between 290 and 420 nm, such that the vitamin may
20 protect hair from photo damage when comprised in a composition of the invention.

Suitable vitamin compounds and their respective derivatives falling within the scope of the present invention include, but are not limited to, the following:

- 16 -

	<u>Vitamin</u>	<u>Derivative</u>
	Vitamin A (Retinol)	Retinyl Palmitate
	Pro Vitamin A (Beta-Carotene)	
	Vitamin B ₁ (Thiamine)	Thiamine Nitrate
5		Thiamine Phosphoric Acid Ester
	Vitamin B ₃ (Niacinamide)	Niacinamide Ascorbate
	Vitamin B ₅ (Pantothenic Acid)	
	Vitamin B ₆ (Pyridoxine)	Pyridoxine Dilaureate
10		Pyridoxine Dioctenoate
		Pyridoxine Dipalmitate
		Pyridoxine Tripalmitate
	Vitamin B ₁₂ (Cyanocobalamin)	
	Vitamin C (Ascorbic Acid)	Ascorbyl Palmitate
15		Ascorbyl Glucoseamine
		Ascorbyl Dipalmitate
		Ascorbyl Stearate
	Vitamin D ₂ (Ergocalciferol)	
	Vitamin D ₃ (Cholecalciferol)	
	Vitamin E (Tocopherol)	Tocopheryl Acetate
20		Tocopheryl Linoleate
		Tocopheryl Nicotinate
		Tocopheryl Succinate
	Vitamin F (Linoleic, Linolenic acids)	
	Vitamin K ₁ (Phylloquinone)	

25

The above-mentioned vitamin compounds and their respective derivatives absorb ultra-violet light within the wavelength region between 290 and 420 nm.

Vitamin compounds and their respective derivatives (including, for example, salt or esterified forms of the vitamin) may be combined and used in the composition of the present invention.

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- 17 -

Examples 6-11, below, demonstrate that applying a composition comprising an anionic hydrolyzed protein comprising sulphur-containing amino acids, a divalent cationic compound and a vitamin compound to the hair provides substantial protection from photo damage, chemical damage and damage caused by grooming.

5 The composition of the present invention comprises from about 1 to 99.89% by weight of a cosmetic carrier. The term "cosmetic carrier" as used herein means the components, other than hydrolyzed protein, divalent cationic compound and vitamin compound, which are generally used in the cosmetic compositions within the scope of the present invention. Such cosmetic compositions may include, but are not
10 limited to, shampoos, conditioners, hair treatment creams, styling gels, mousse, pump hair sprays and aerosol hair sprays and foams, set lotions, blow styling lotions, hair color lotions, hair relaxing compositions, permanent wave first agents, and permanent wave second agents.

 Specifically, cosmetic carriers, as defined herein, include, but are not
15 limited to, water, aqueous solutions, detergents, emollients, surfactants, foam boosters, thickeners, fatty esters, ethers, alcohols, polymers, preservatives, color, dyes, fragrance and other ingredients known to those skilled in the art.

 Various specific examples of compositions of the present invention were tested for beautifying effects and performance by skilled hairdressers. Based on such
20 tests, the compositions were found to impart an excellent look, smoothness, feel and shine to the hair, as well as to provide excellent comability under wet and dry conditions. It was also observed that the compositions of the present invention made the hair more manageable, indicating that such compositions were effective as moisturizing, softening and bodifying treatments.

25 Hair may be treated with the compositions of the present invention by applying effective amounts to the hair. When the composition of the present invention is present in the form of a shampoo or conditioner, the application is usually followed by rinsing the hair with water and finally drying the hair. When the composition of the

- 18 -

present invention is a conditioner, treatment cream, styling gel or mousse, it may alternatively be retained in the hair or rinsed.

Further, the compositions of the present invention may be separated and applied to the hair in effective amounts by applying such amounts in multiple steps to treat the hair as may be practiced with such products as perm waves, hair straighteners and hair relaxers. For example, the protein and divalent cationic compound may be in the waving solution and the vitamin compound applied as a separate step before neutralizing the hair.

Many features of the invention will become apparent in the course of the following description of the exemplary embodiments which are given for illustration of the invention and are not intended to be limiting thereof.

The term "polyamino cationic agent," as recited in the specification and claims, is defined to include within the scope of that term any positively charged compound having a plurality of positively charged amino groups including, but not limited to, the compounds referenced in the specification and claims.

EXAMPLE 1

Effective Molar Ratios of Divalent Cationic Compound to Anionic Amino Acids in the Hydrolyzed Protein

Bleached hair swatches were colored with an oxidation dye, o-Cl-p-Phenylenediamine sulfate/m-Phenylenediamine("o-Cl-PPD/MPD"), and then treated with one of the following solutions for 10 minutes at 40°C:

- a) water
- b) 3% by weight of hydrolyzed protein from wheat (Cropeptide W);
- c) 3% by weight of hydrolyzed protein from wheat (Cropeptide W), wherein the molar ratio of the divalent cationic compound to anionic acidic amino acids in the hydrolyzed protein was either 0.1:1, 0.5:1, 1.0:1, 2.0:1 or 4.0:1.

In addition, bleached hair swatches were colored with an oxidation dye, o-Chloro-p-Phenylenediamine sulfate/4-Ethoxy-m-phenylenediamine sulfate ("o-Cl-

- 19 -

PPD/DAPS") and then treated with one of the following solutions for 10 minutes at 40°C:

- a) water
- b) 4% by weight of hydrolyzed protein from milk

5 (Hydrolactin);

- c) 4% by weight of hydrolyzed protein from milk

(Hydrolactin), wherein the molar ratio of the divalent mineral to anionic acidic amino acids in the hydrolyzed protein was either 0.1:1, 0.5:1, 1.0:1, 2.0:1 or 4.0:1.

The divalent cationic compounds used were Mg^{++} , Zn^{++} , and Ca^{++} from the salts gluconate and pantothenate. The molar ratios of divalent cationic compounds to anionic amino acids of the hydrolyzed protein were based on the total number of moles of the salt to the total number of moles of the anionic amino acids of the hydrolyzed protein used in the solution. The molar ratios of divalent cationic compounds to anionic amino acids of the hydrolyzed protein set forth in the remaining examples were calculated in the same manner as was described above.

The o-Cl-PPD/MPD-treated hair swatches in the above-mentioned solutions were irradiated in a Weather-Ometer, which is a simulated sunlight exposure machine using a xenon arc lamp, for 5 hours. The o-Cl-PPD/DAPSD treated hair swatches in the above-mentioned solutions were exposed in the Weather-Ometer for 20 2 hours. The loss of "lightness" (change in L value) and blue color (change in b value) in the treated hair swatches were then determined according to the CIE L*A*B color difference system.

Changes in both L and b values represent color loss due to the irradiation treatment. The results are shown in Figures 1a-d. The effective molar ratio range of divalent cationic compounds to anionic amino acids of the hydrolyzed protein that protect the hair from photo damage was from about 0.2:1 to 4.0:1, preferably from about 0.5:1 to 1.0:1, and most preferably at about 0.5:1. Also, compared to hair treated with the hydrolyzed protein only, hair treated with the aforementioned molar ratios of divalent cationic compounds to anionic amino acids showed significantly less color loss.

- 20 -

EXAMPLE 2**Mechanical Strength of Irradiated Hair Treated
With Hydrolyzed Protein Alone or in Combination
With Divalent Cationic Compounds**

5 Samples of normal brown hair were treated for 10 minutes at 40°C with solutions containing either 25% by weight of hydrolyzed protein from wheat (Cropeptide W) or 25% by weight of hydrolyzed protein from wheat (Cropeptide W) with the divalent cationic compound zinc gluconate, wherein the molar ratio of divalent cationic compound to anionic amino acids of the hydrolyzed protein was about 0.5:1.

10 Individual samples of treated hair were irradiated in the Weather-Ometer for either 2 hours or 4 hours and then equilibrated at 25°C at 50% relative humidity for 24 hours. The treated hair was then stretched by the Instron Tester under the same conditions. The Fc value was obtained from the stress/strain curve and represents the condition of the hair. The higher the Fc value, the stronger the hair. This technique is
15 described in United States Patent Application Serial No. 08/218,565, filed March 28, 1994, and is hereby incorporated by reference in its entirety.

 The results are shown in Figure 2. Hair treated with the hydrolyzed protein and divalent cationic compound was significantly stronger than hair treated with hydrolyzed protein only, and stronger than the untreated hair. This increase in strength
20 suggests that there may be a synergistic effect between protein and the divalent cationic compound in protecting hair from photo damage.

EXAMPLE 3**Effect of Chlorinated Water on Hair Treated
25 With Hydrolyzed Protein Alone or in Combination
 With Divalent Cationic Compounds**

 Samples of normal brown hair were treated for 10 minutes at 40°C with solutions containing either 15% by weight of hydrolyzed protein (Cropeptide W) alone or
30 15% by weight of hydrolyzed protein (Cropeptide W) with divalent cationic minerals from the salts magnesium gluconate, zinc gluconate and calcium pantothenate. The

- 21 -

molar ratio of divalent cationic minerals to anionic amino acids of the hydrolyzed protein was about 0.5:1. This ratio was calculated in the same manner as was described in Example 1.

5 Samples of treated hair were subjected to the following process up to three times (i.e., three cycles): (a) protein treated; (b) rinsed with water; (c) placed in a 1% NaOCl solution for 10 minutes at room temperature; (d) shampooed using 10% ammonium lauryl sulfate ("ALS") solution; and (e) rinsed with water. Protein loss from the hair was then determined according to the procedure described in the published procedure "A Simple and Sensitive Technique, Based on Protein Loss Measurement, to
10 Assess Surface Damage to Human Hair," Journal of the Society of Cosmetic Chemists, Vol. 44, pp. 163-175, (1993).

The results are shown in Figure 3, indicating that the protein loss due to chlorinated water was less in the hair treated with hydrolyzed protein and divalent cationic compounds than in the hair treated with hydrolyzed protein only. This decrease
15 in protein loss suggests that there may be a synergistic effect between the hydrolyzed protein and divalent cationic compounds in protecting hair from oxidative damage caused by chlorinated water.

EXAMPLE 4

20 **Mechanical Strength of Hair Treated With
Either Hydrolyzed Protein or Hydrolyzed Protein
With Divalent Cationic Compounds
After Extended Periods of Irradiation**

25 Normal brown hair was treated for 10 minutes at 40°C with either a solution containing 25% by weight of hydrolyzed protein from wheat (Cropeptide W) or 25% by weight of hydrolyzed protein from wheat (Cropeptide W) together with divalent cationic minerals from the salts magnesium gluconate, zinc gluconate, and calcium pantothenate. The molar ratio of divalent cationic compound to anionic amino acids of
30 the hydrolyzed protein was about 0.5:1.

- 22 -

Samples of the treated hair were irradiated in the Weather-Ometer from 8 hours up to three days. The irradiated hair was then equilibrated at 25°C at 50% relative humidity for 24 hours and then stretched by the Instron Tester under the same conditions. The Fc value was obtained from the stress/strain curve and represents the condition of the hair, as described in Example 2.

The results are shown in Figure 4. The addition of divalent cationic compounds to the hydrolyzed protein solution significantly increased the mechanical strength of the hair, suggesting the presence of a synergistic effect between the hydrolyzed protein and divalent cationic compounds in creating a long lasting protective effect against photo damage.

EXAMPLE 5

Comparison Of The Protective Efficacy of Hydrolyzed Protein From Wheat and Hydrolyzed Protein From Collagen Which Was Either Succinylated or Sulfonated

Normal brown hair was individually treated with the following: a) a solution containing 24% by weight of hydrolyzed protein from wheat (Cropeptide W) and divalent cationic compounds from the salts zinc gluconate, magnesium gluconate, and calcium pantothenate; b) a solution containing 25% by weight of succinylated hydrolyzed protein from collagen and divalent cationic compounds from zinc gluconate, magnesium gluconate, and calcium pantothenate; and c) a solution containing 30% by weight of sulfonated hydrolyzed protein from collagen and divalent cationic compounds from the salts zinc gluconate, magnesium gluconate, and calcium pantothenate. The molar ratio of the divalent cationic compounds to the anionic amino acids in the above-mentioned proteins was about 0.2:1. The average molecular weight of succinylated or sulfonated proteins suitable for the present invention may be from 500 up to 500,000 daltons, or more.

The treated hair was irradiated in the Weather-Ometer for 4 hours and then tested by the Instron at 25°C, 50% relative humidity. Based on the stress-strain curve, the condition of the hair was determined according to the Fc values, as described in

- 23 -

Example 2. The results are shown in Figure 5. Compared to the hydrolyzed protein from wheat, the succinylated and sulfonated hydrolyzed protein from collagen both significantly improved the condition of the hair.

5

EXAMPLE 6

Effect of Combining Vitamin Compounds With A Hydrolyzed Protein and Divalent Cationic Compounds In Protecting Hair From Photo damage

10

Bleached hair swatches were colored with an oxidation dye (o-Cl-PPD/MPD) and then treated for 10 minutes at 40°C with solutions containing 3% hydrolyzed protein from wheat (Cropeptide W) and 1.8% divalent minerals (Mg Gluconate 0.6%, Zn Gluconate 0.6%, Ca Pantothenate 0.6%). The hair swatches were treated with the above-mentioned solutions with and without 5% Vitamin A or 5% Vitamin E. The molar ratio of the divalent cationic compounds to the anionic amino acids in the hydrolyzed protein in these above-mentioned compositions was about 0.5:1.

15

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The treated hair swatches were irradiated in the Weather-Ometer for 2 hours, and the loss of color (change in the b value) and lightness (change in the L value) was determined according to the CIE L*A*B color difference system. Minimum change in the b value or the L value represents negligible color loss due to the irradiation treatment. The results shown below in Table 2 indicate that adding Vitamin A or Vitamin E to the hydrolyzed protein and divalent cationic compound mixture provided further protection against color loss.

- 24 -

TABLE 2**Color Loss Of Irradiated Hair**

Treatment	Change In b (After 2 Hours)	Change In b (After 4 Hours)	Change In L value (After 2 hours)
Protein + Minerals	--	--	51.1
Protein + Minerals + Vitamin A	--	--	42.5
Protein + Minerals	6.12	9.49	--
Protein + Minerals + Vitamin E	0.85	1.82	--

EXAMPLE 7**Effect of Irradiating Hair Treated With Hydrolyzed Protein,
Divalent Cationic Compounds and Vitamin Compounds**

Normal brown hair was treated for 10 minutes at 40°C with the following solutions listed in Table 3:

TABLE 3

Hydrolyzed Protein from Wheat	Minerals	Vitamins	% Residual Cystine	Molar Ratio of Divalent Cationic Compounds to Anionic Amino Acids in Hydrolyzed Protein
(1) 24.0 g	---	---	1.40	
(2) ---	---	5.0 g (Vitamin E)		
(3) 23.0 g	5.1 g (1.7 g Zn Gluconate 1.7 g Mg Gluconate 1.7 g Ca Pantothenate)	---	1.34	0.6:1
(4) 7.2 g	5.1 g (1.7 g Zn Gluconate 1.7 g Mg Gluconate 1.7 g Ca Pantothenate)	5.0 g (1.25 g Vitamin A 1.25 g Vitamin E 1.25 g Vitamin B1 1.25 g Vitamin B6)	0.42	0.6:1

- 25 -

The pH range of the compositions described in Table 3 was from about 4 to 6.

The treated hair was irradiated in the Weather-Ometer for 24 hours. The irradiated hair samples were tested by the Instron Tester at 23 °C, 50% relative humidity.

5 Based on the stress/strain curves, the Condition Factor (Fc) values were determined using the same procedure described in Example 2 and the percent damage calculated.

The results are shown in Figure 6. The hydrolyzed protein from wheat (average molecular weight 2500 daltons, molar ratio of anionic to cationic amino acids at about 13.9:1.0, cystine content about 1.40%) provided significant protection from photo
10 damage to the hair.

Further, addition of divalent cationic compounds to the protein solution at about 0.6:1 molar ratio of cationic to anionic amino acids in the hydrolyzed protein (solution (3)) further decreased the damage despite the fact that the protein had a lower protein content (23 grams) and a lower cystine content, 1.34%. Such data is consistent
15 with the theory discussed herein that the hydrolyzed protein is bound to the hair via divalent cationic compound bridges.

Moreover, in the solution (4), wherein the protein content is only 7.2 grams and the protein comprises only 0.42% cystine, an even greater decrease in photo damage was obtained when vitamin compounds were added to the divalent cationic
20 compound/hydrolyzed protein solution, indicating a synergistic effect in combining a divalent cationic compound/hydrolyzed protein with a vitamin compound. This would suggest that compositions comprising hydrolyzed protein, divalent cationic compounds, and vitamins in the proportions according to the present invention impart excellent protection to hair against photo damage.

25

EXAMPLE 8

Efficacy of Hydrolyzed Protein, Divalent Cationic Compounds and Vitamin Compounds In Retarding Protein Loss From Hair

30

In reference to the shampoo, conditioner and treatment composition described in Table 4, normal brown hair was washed with shampoo for 5 minutes at room

- 26 -

temperature, then treated with conditioner for 5 minutes at room temperature. The treatment composition was applied to the conditioned hair for 10 minutes at 40°C, and then placed in a 1% NaOCl solution for 10 minutes at room temperature. Samples of the treated hair were subjected to the shampoo, conditioner, treatment, and NaOCl steps up to 5 8 times (i.e., 8 cycles). Protein loss from the hair was determined after 2,4 and 8 cycles using the procedure described in Example 3.

- 27 -

TABLE 4

CETA Names	Shampoo	Conditioner	Treatment
Self Emulsifying Glyceryl Ester		6.0%	6.0%
Cetrimonium Chloride		3.5%	3.5%
Dicetyldimonium Chloride		3.0%	3.0%
Hydrolyzed protein from wheat	0.1%	0.1%	3.0%
Cetearyl Alcohol		2.0%	2.0%
Trimethylsilylamodimethicone		0.7%	0.7%
Divalent Minerals (Total)	(0.03%)	(0.03%)	(0.75%)
Mg Gluconate	0.01%	0.01%	0.25%
Zn Gluconate	0.01%	0.01%	0.25%
Ca Pantothenate	0.01%	0.01%	0.25%
Vitamins (Total)	(0.02%)	(0.02%)	(0.22%)
Thiamine Hydrochloride			0.1%
Pyridoxine Hydrochloride			0.1%
Tocopheryl Acetate	0.01%	0.01%	0.01%
Retinyl Palmitate	0.01%	0.01%	0.01%
Menthol			0.1%
Phytolipid and Hyaluronic Acid			0.1%
Apricot Seed (Apricot Kernel Powder produced by Alban Muellen, Inc. of Paris, France)			0.25%
Ammonium Lauryl Sulfate	25%		
Cocamidopropyl Betaine	10%		
Sodium Lauroyl Sarcosinate	5%		
Pearlizing Agent	0.8%		
Methyl Gluceth-20	0.25%		
Polyquaternium-4	0.1%		

The molar ratios of divalent cationic compounds to anionic amino acids of the hydrolyzed protein in both the shampoo and conditioner were about 1:1, and the ratio

- 28 -

in the treatment composition was about 0.8:1. The pH range of the above-mentioned compositions was from about 4 to 6.

The results are shown in Figure 7. After 8 cycles, hair treated with the hydrolyzed protein, divalent cationic compound and vitamin compound mixture exhibited about 25% less protein loss due to chlorinated water than the untreated hair.

EXAMPLE 9

Mechanical Strength of Irradiated Hair Treated With Hydrolyzed Protein, Divalent Cationic Compounds and Vitamin Compounds

Normal brown hair was treated with the treatment cream described below and irradiated in the Weather-Ometer for 2 hours. The treatment and irradiation steps were repeated up to 7 times (i.e., 7 cycles). After each cycle, the Fc values of the irradiated hair were determined by the Instron Tester, as described in Example 2, at 23 °C, 50% relative humidity.

The treatment cream contained the following: Self Emulsifying Glyceryl Ester 6.0%, Cetrimonium Chloride 3.5%, Dicetyldimonium Chloride 3.0%, Hydrolyzed protein from wheat 3.0%, Cetearyl Alcohol 2%, Trimethylsilylamodimethicone 1.0%, Divalent Minerals 1.5% (Mg Gluconate 0.5%, Zn Gluconate 0.5%, and Ca Pantothenate 0.5%), Vitamin Compounds 0.81% (Thiamine Hydrochloride 0.1%, Pyridoxine Hydrochloride 0.5%, Tocopheryl Acetate 0.2%, Retinyl Palmitate 0.01%), Menthol 0.1%, Phytolipid and Hyaluronic Acid 0.1%. The molar ratio of the divalent cationic compounds to anionic amino acids in the hydrolyzed protein was about 1.6:1. The pH range of the composition was from about 4 to 6.

The results are shown in Figure 8, and indicate that, after each cycle, the condition of the treated hair, as reflected by the higher Fc values, was found to be significantly better than the condition of untreated hair. From this data, it was observed that treating hair with a treatment cream comprising hydrolyzed protein, divalent cationic compounds and vitamin compounds provided substantial protection to the hair from photo damage.

- 29 -

EXAMPLE 10**Efficacy of Hydrolyzed Protein, Divalent Cationic Compounds and Vitamin Compounds In Protecting Hair From Blow Drying and Curling Iron**

5

In reference to the shampoo, conditioner and treatment composition described in Table 4, normal brown hair was washed with the shampoo for 5 minutes at room temperature, then treated with the conditioner for 5 minutes at room temperature daily. The treatment composition was applied to the conditioned hair for 10 minutes at 40°C twice a week. Each day, the hair was blow-dried for 5 minutes and then pressed through a curling iron 50 times. This regimen was carried out for two weeks. The mechanical property of the hair (treated and untreated) was tested by the Instron Tester at 23 C, 50% relative humidity according to the procedure described in Example 2.

10

The molar ratios of divalent cationic compounds to anionic amino acids of the hydrolyzed protein in both the shampoo and conditioner were about 1:1, and the ratio in the treatment composition was about 0.8:1. The pH range of the above-mentioned compositions was between about 4 and 6.

15

It was observed that the energy required to break the hair treated with the above-described compositions after the first week was 8% higher than that required to break untreated hair, and 11% higher after two weeks. This would indicate that compositions comprising hydrolyzed protein, divalent cationic compounds, and vitamin compounds in the proportions according to the present invention impart greater protection to the hair against excessive heat by grooming devices, such as blow dryers, curling irons and the like.

20

25

EXAMPLE 11**Mechanical Strength of Irradiated and Combed Normal, Bleached and Permed Hair Treated With Hydrolyzed Protein, Divalent Cationic Compounds and Vitamin Compounds**

30

In reference to the shampoo, conditioner and treatment in Table 4, bleached hair, permed hair and virgin brown hair were treated with the shampoo for five

- 30 -

minutes at room temperature, rinsed with water, treated with the conditioner for 5 minutes at room temperature, rinsed with water and then treated with the hair treatment for 10 minutes at 40°C. The above-mentioned hair samples were treated in this manner twice a week for 14 days. During this time period, the treated hair was irradiated in the Weather-Ometer each day for four hours. Such exposure in the Weather-Ometer is equivalent to 1 day in the sun. Combing (200 strokes) was done on the wet treated hair each day after irradiation. The irradiated and combed hair was tested by the Instron Tester at 23°C, 50% relative humidity, as described in Example 2, to ascertain the Fc value (i.e., mechanical strength of the hair).

The results are shown in Figure 9. The higher Fc values indicate that the condition of the hair treated with the shampoo, conditioner and treatment, as described in Table 4, was stronger and in better condition than the untreated hair. This suggests that the hydrolyzed protein/divalent cationic compound/vitamin compound mixture is effective in protecting bleached hair, permed hair, and normal from excessive combing and exposure to sunlight.

Unless otherwise indicated, all weight percentages set forth above refer to the weight percent of the compound based on the total weight of the composition.

Although this invention has been described with reference to specific examples, it will be apparent to one skilled in the art, that various modifications may be made thereto which fall within the scope and preview of the invention.

Various publications are cited herein which are hereby incorporated by reference in their entirety.

EXAMPLE 12

Effective Molar Ratios of Polyamino Cationic Agents to Anionic Amino Acids of the Hydrolyzed Protein

Figure 10 shows test results measuring color loss in untreated hair and hair treated with compositions comprising a negatively charged hydrolyzed protein and polyamino cationic agent in various molar ratios. The testing reflected in Figure 10 was carried out as follows:

- 31 -

(1) Bleached hair swatches were colored with an oxidation dye, o-Cl-p-Phenylenediamine sulfate/m-Phenylenediamine("o-Cl-PPD/MPD"), and then treated with one of the following solutions for 10 minutes at 40°C:

- 5 (a) 3% by weight of hydrolyzed protein from soy Hydrosol
2000 SF;
- (b) 3% by weight of hydrolyzed protein from soy Hydrosol
2000 SF and cationic amino acid compound. The molar ratio of the cationic amino acid
compound to anionic amino acids in the hydrolyzed protein was either 0.1:1, 0.5:1, 1.0:1,
2.0:1 or 4.0:1. Stated another way, the ratio of the amino groups of the cationic amino
10 acid compound to anionic amino acids of the hydrolyzed protein was either 0.2:1, 1.0:1,
2.0:1, 4.0:1 or 8.0:1; and
- (c) water containing an amount of cationic amino acid
compound corresponding to that used in each of the solutions described in (b) above.

 The cationic amino acid compound used in these tests was arginine
15 hydrochloride (Brand name: L-Arginine Hydrochloride sold by Ajinomoto of Japan).
The molar ratio of the cationic amino acid compound to anionic amino acids of the
hydrolyzed protein were based on the total number of moles of the arginine to the total
number of moles of the anionic amino acids of the hydrolyzed protein used in the
solution.

20 The o-Cl-PPD/MPD-treated hair swatches in the above-mentioned
solutions were irradiated in a Weather-Ometer, which is a simulated sunlight exposure
machine using a xenon arc lamp, for 5 hours. The loss of "lightness" (change in L value)
in the treated hair swatches were then determined according to the CIE L*A*B* color
difference between the change in the L value of treated hair and non-treated hair (i.e., hair
25 treated with water alone).

 Percentage of improvement represents changes in L which reflects color
loss due to the irradiation treatment. Compared to hair treated with the hydrolyzed
protein and dibasic amino acid treated with the aforementioned molar ratios of dibasic
amino acid compounds to anionic amino acids showed significantly less color loss.

- 32 -

The expected value was determined by adding the separate effects of treating hair with protein and dibasic amino acid compounds alone.

The experimental data in Figure 10 demonstrate that treatment of hair with a composition comprising the molar ratios of cationic amino acid compounds and hydrolyzed protein yields an unexpected synergistic protective effect in protecting hair.

Based on the experimental data in Figure 10, treatment of hair with hydrolyzed protein alone resulted in modest protection against photo damage, whereas treatment of hair with a polyamino cationic agent alone yielded minimal or no protection. It was therefore expected that combining the hydrolyzed protein with the polyamino cationic agent would provide little or no additional advantages beyond those obtained with protein alone.

However, as depicted in Figure 10, the level of protection obtained by treating hair with both protein and polyamino cationic agents in the molar ratios specified in the claims was greater than that achieved when hair was treated with either of the two components alone, and also greater than the expected protection from the additive combination of the hydrolyzed protein with the polyamino cationic agent.

The data in Figure 10 provides a reasonable indication that the polyamino cationic agents and hydrolyzed proteins disclosed in the specification, when combined in the specified ratios, would provide the same or similar additive effects in protecting hair as the tested polyamino cationic agent and hydrolyzed protein.

- 33 -

EXAMPLE 13**Example of Hair Treatment Compositions Comprising
Hydrolyzed Protein and Polyamino Cationic Agents****TABLE 5****SHAMPOO**

CFTA Names	%W/W
Water	67.700
Hydroxypropyl Methylcellulose	0.400
Methylparaben	0.100
Ethylparaben	0.100
Disodium EDTA	0.100
Taurine	0.010
Trehalose	0.001
Arginine HCl	0.010
D&C Red	>0.001
FD&C Red	>0.001
Lauryldimonium Hydroxypropyl Hydrolyzed Soy Protein	0.010
Hydrolyzed Soy Protein and Wheat Amino Acid	0.100
TEA-Lauryl Sulfate	23.000
Cocamide MIPA	5.000
SLES and Glycol Distearate and Cocamide MEA, Laureth 10	1.500
Na Citrate	0.500
Citric Acid	0.400
Phenoxyethanol	0.500
Fragrance	0.500

- 34 -

TABLE 6**PROTEIN HAIR TREATMENT**

CFTA Name	% W/W
Water	68.00
Hydroxy Ethcellulose	0.800
Methyl Paraben	0.200
Ethyl Paraben	0.100
Disodium EDTA	0.100
Arginine Hydrochloride	3g
Taurine	0.100
Trehalose	0.001
Hydrolysed Soy Protein	25.00
Lauryldimonium Hydroxypropyl Hydrolyzed Soy Protein	0.010
Hydrolysed Soy Protein and Wheat Amino Acid	0.1000
Polyquaternium II	0.5000
Cetrimonium Chloride	1.000
Phenoxyethanol	0.700
Laureth-23	0.150
Phosphoric Acid	0.200

In these compositions, the polyamino cationic agent is arginine hydrochloride. The ratio of the arginine hydrochloride to anionic amino acids of the hydrolyzed protein in the above shampoo and protein hair treatment were about .6:1 to .9:1. The pH range of these compositions was from about 4 to 6. The above compositions would provide a greater additive effect in protecting hair than the combined individual effects of the same or similar compositions containing the hydrolyzed protein or polyamino cationic agent alone.

- 35 -

Various publications are cited herein, the contents of which are hereby incorporated by reference in their entirety.

- 36 -

Claims

1. A cosmetic composition for treating hair comprising:
 - a. from about 0.1 to 50% by weight of a hydrolyzed protein
5 comprising anionic and cationic amino acids, wherein the molar ratio of the anionic amino acids to cationic amino acids of the hydrolyzed protein is at least 2.1:1.0, wherein the hydrolyzed protein comprises at least 0.25% by weight of a sulphur-containing amino acid, and wherein the hydrolyzed protein has an average molecular weight of less than 500,000 daltons;
 - 10 b. a polyamino cationic agent, wherein the polyamino cationic agent is a cationic amino acid compound having a plurality of amino groups, wherein said amino acid compound is selected from the group consisting of histidine, arginine, lysine, hydroxylysine, ornithine or combinations thereof, wherein the ratio of the amino groups of the cationic amino acid compound to the anionic amino acids of the hydrolyzed protein
15 is from about 0.2:1 to 2:1; and
 - c. from about 1 to 99.9% by weight of cosmetic carrier.
2. A cosmetic composition for treating hair comprising:
 - a. from about 0.1 to 50% by weight of a hydrolyzed protein
20 comprising anionic and cationic amino acids, wherein the molar ratio of the anionic amino acids to cationic amino acids of the hydrolyzed protein is at least 2.1:1.0, wherein the hydrolyzed protein comprises at least 0.25% by weight of a sulphur-containing amino acid, and wherein the hydrolyzed protein has an average molecular weight of less than 500,000 daltons;
 - 25 b. a polyamino cationic agent, wherein the cationic polyamino agent is a peptide, wherein at least 2 of the amino acids of the peptide are cationic amino acid compounds selected from the group consisting of histidine, arginine, lysine, hydroxylysine, ornithine or combinations thereof; wherein the ratio of the amino groups of the peptide to the anionic amino acids of the hydrolyzed protein is from about 0.2:1 to
30 2:1; and

- 37 -

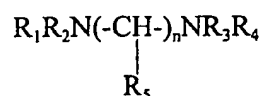
c. from about 1 to 99.9% by weight of cosmetic carrier.

3. A cosmetic composition for treating hair comprising:

a. from about 0.1 to 50% by weight of a hydrolyzed protein

5 comprising anionic and cationic amino acids, wherein the molar ratio of the anionic amino acids to cationic amino acids of the hydrolyzed protein is at least 1.1:1.0, wherein the hydrolyzed protein comprises at least 0.25% by weight of a sulphur-containing amino acid, and wherein the hydrolyzed protein has an average molecular weight of less than 500,000 daltons;

10 b. a polyamino cationic agent having the formula:

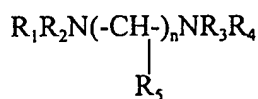


15 where $n=1-6$ and R_1, R_2, R_3, R_4 , and R_5 are selected from the group consisting of H, an alkyl group or compound comprising one or more amino groups, wherein the molar ratio of the amino groups of the polyamino cationic agent to the anionic amino acids of the hydrolyzed protein is from about 0.2:1 to 2:1;

c. from about 1 to 99.9% by weight of cosmetic carrier.

20

4. The hair treatment composition according to claim 3, wherein the polyamino cationic agent is a diamino compound having the formula:



25

where $n=1-6$ and R_1, R_2, R_3, R_4 , and R_5 are selected from the group consisting of H, or an alkyl group.

30

5. The hair treatment composition according to claim 1, 2, 3 or 4 wherein the molar ratio of the anionic amino acids to cationic amino acids of the hydrolyzed protein is from about 2.1:1 to 15:1.

- 38 -

6. The hair treatment composition according to claim 1, 2, 3, or 4 wherein the ratio of the amino groups of the polyamino cationic agent to the anionic amino acids of the hydrolyzed protein is from about 1:1 to 2:1.

5 7. The hair treatment composition according to claim 1, 2, 3, or 4 wherein the ratio of the amino groups of the polyamino cationic agent to anionic amino acids of the hydrolyzed protein is about 1.0:1.

8. The hair treatment composition according to claim 1, 2, 3, or 4 wherein
10 the source of the hydrolyzed protein is selected from the group consisting of wheat, milk, keratin, soy, collagen or mixtures thereof.

9. The hair treatment composition according to claim 1, 2, 3, or 4 further
15 comprising 0.01 to about 5% by weight of a vitamin compound selected from the group consisting of vitamins having antioxidant properties and vitamins which absorb ultraviolet light within the wavelength region between 290 nm and 420 nm.

10. The hair treatment composition according to claim 9, wherein the
20 vitamin compound is selected from the group consisting of: (1) Vitamin A, an esterified form of Vitamin A and salt form of Vitamin A; (2) Vitamin B₁, an esterified form of Vitamin B₁ and a salt form of Vitamin B₁; (3) Vitamin B₆, an esterified form of Vitamin B₆ and a salt form of Vitamin B₆; (4) Vitamin E, an esterified form of Vitamin E and a salt form of Vitamin E; (5) Beta-Carotene, an esterified form of Beta-Carotene and a salt form of Beta-Carotene; (6) Vitamin B₃, Vitamin B₅, Vitamin B₁₂, Vitamin C, Vitamin D₂,
25 Vitamin D₃, Vitamin F, Vitamin K₁, and their esterified forms and salt forms; and (7) Retinyl Palmitate, Thiamine Nitrate, Thiamine Phosphoric Acid Ester, Niacinamide Ascorbate, Pyridoxine Dilaureate, Pyridoxine Dioctenoate, Pyridoxine Dipalmitate, Pyridoxine Tripalmitate, Ascorbyl Palmitate, Ascorbyl Glucoseamine, Ascorbyl Dipalmitate, Ascorbyl Stearate, Tocopheryl Acetate, Tocopheryl Linoleate, Tocopheryl
30 Nicotinate, Tocopheryl Succinate.

- 39 -

11. The hair treatment composition according to claim 1, wherein the polyamino cationic acid compound comprises a salt selected from the group consisting of hydrochloride, hydrochloride monohydrate, monohydrochloride monohydrate, methyl ester dihydrochloride, ethyl ester dihydrochloride, dihydrochloride, or combinations thereof.

12. The hair treatment composition according to claim 2, wherein the peptide consists essentially of 2 amino acids, where said 2 amino acids are cationic amino acid compounds.

13. The hair treatment composition according to claim 12, wherein the cationic amino acid compounds are selected from the group consisting of histidine, arginine, lysine, hydroxylysine, ornithine or any combination thereof.

14. The hair treatment composition according to claim 1, 2, 3, or 4 wherein the pH of the composition is about 3.5 to 10.

15. A method for treating hair comprising applying to the hair the compositions of claims 1, 2, 3 or 4.

1 / 11

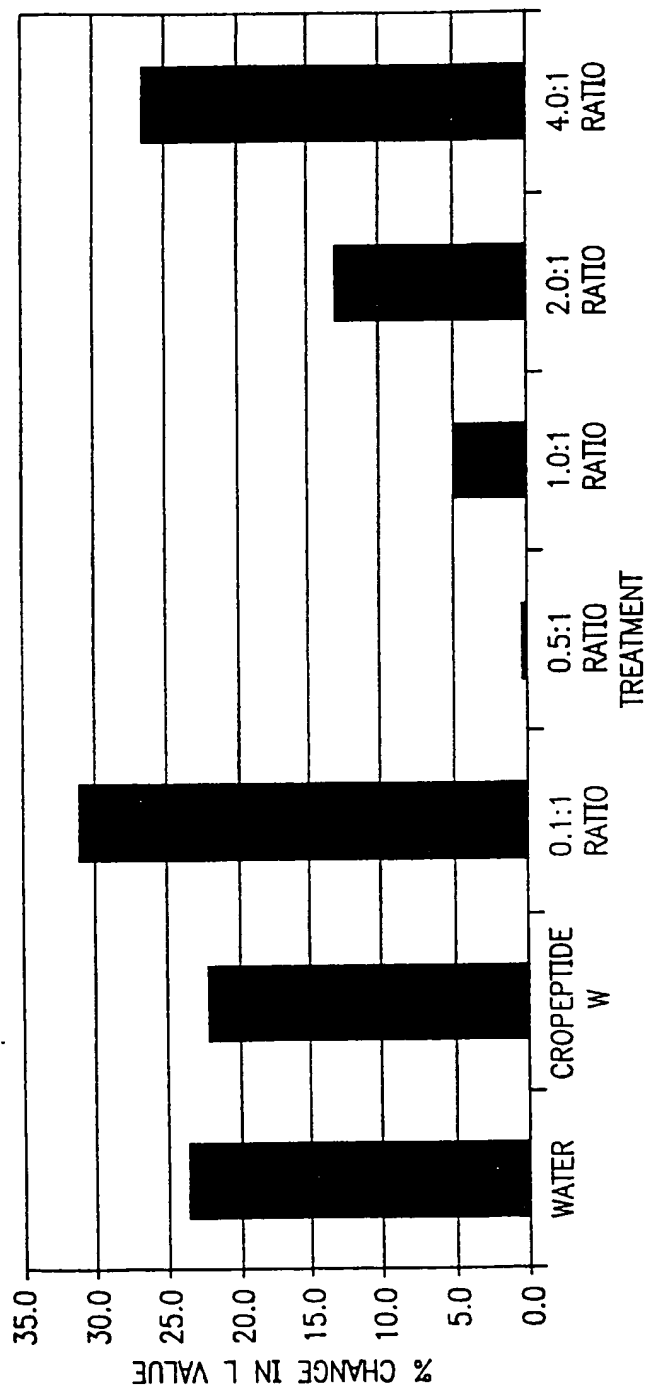


FIG. 1A

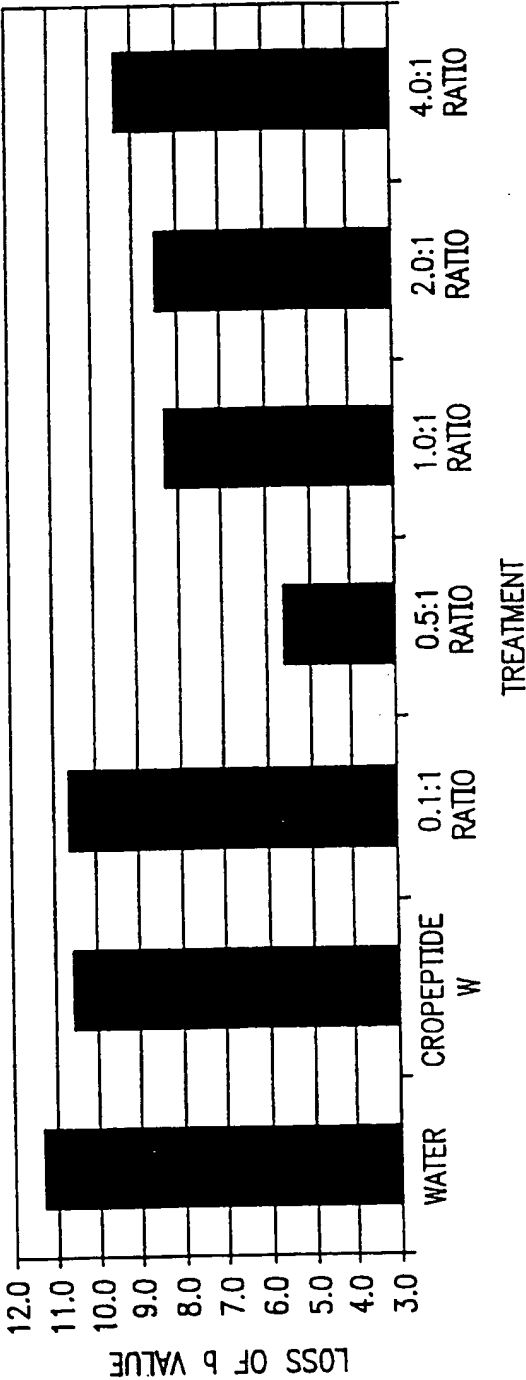


FIG. 1B

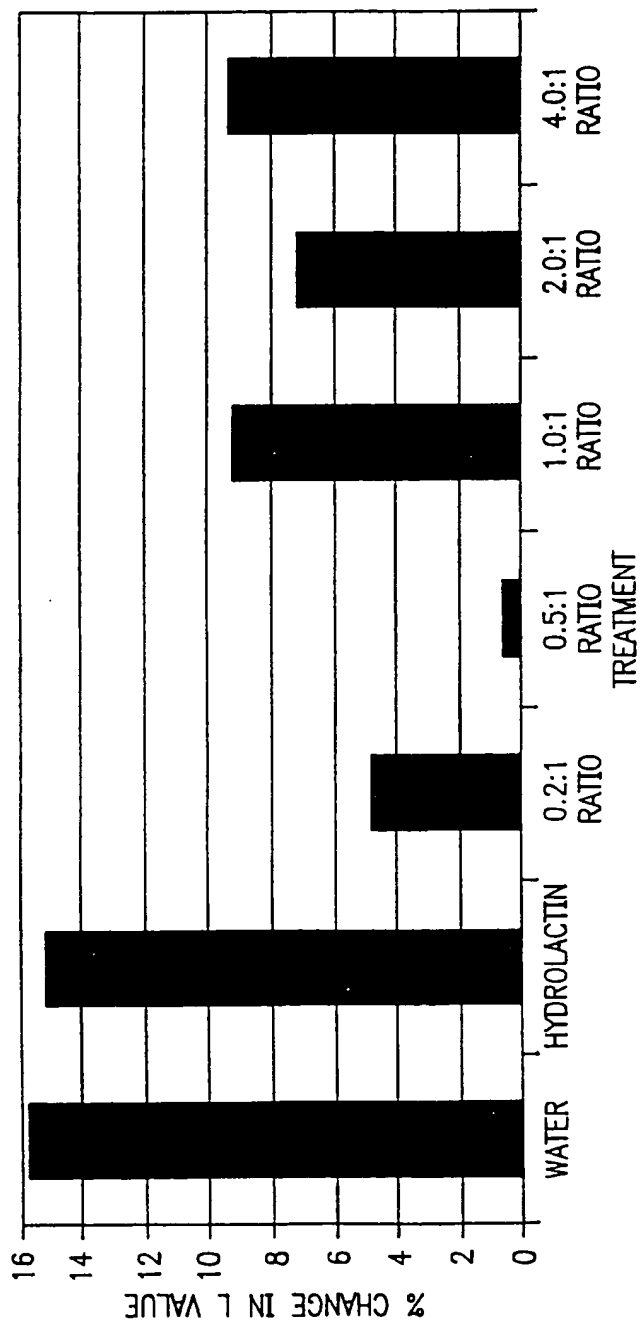
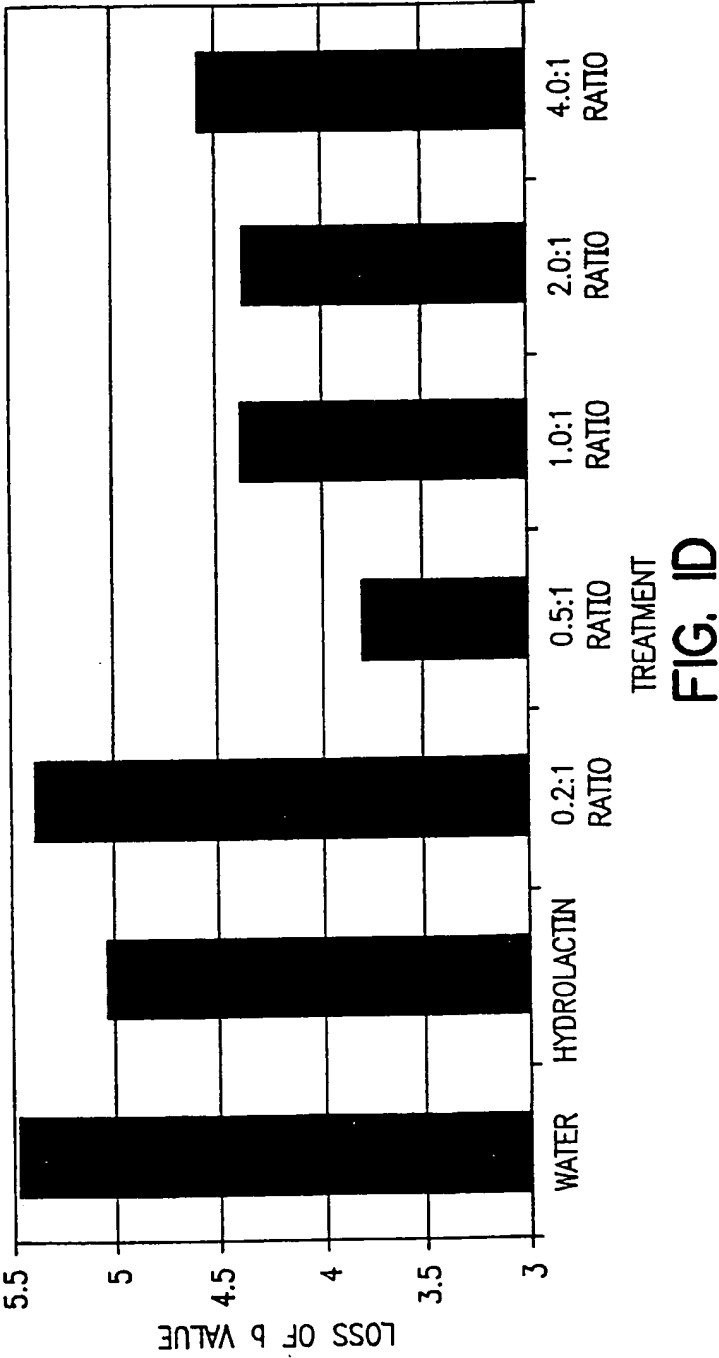


FIG. 1C



5/11

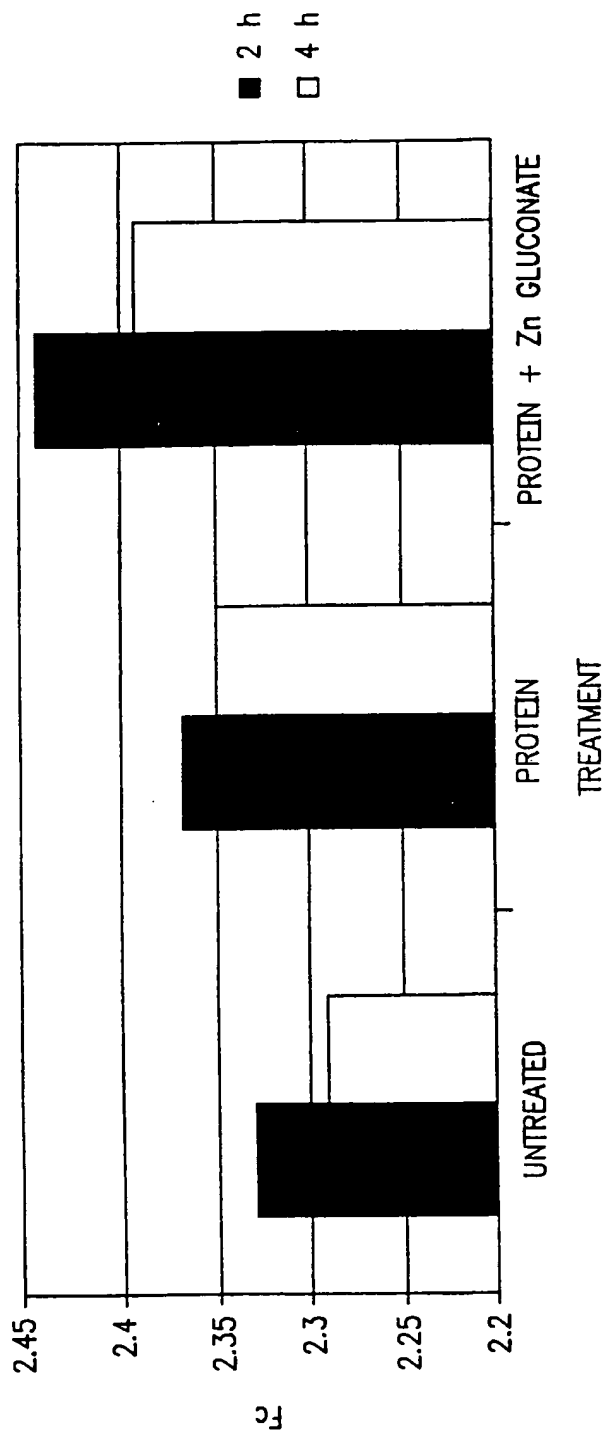


FIG. 2

6 / 11

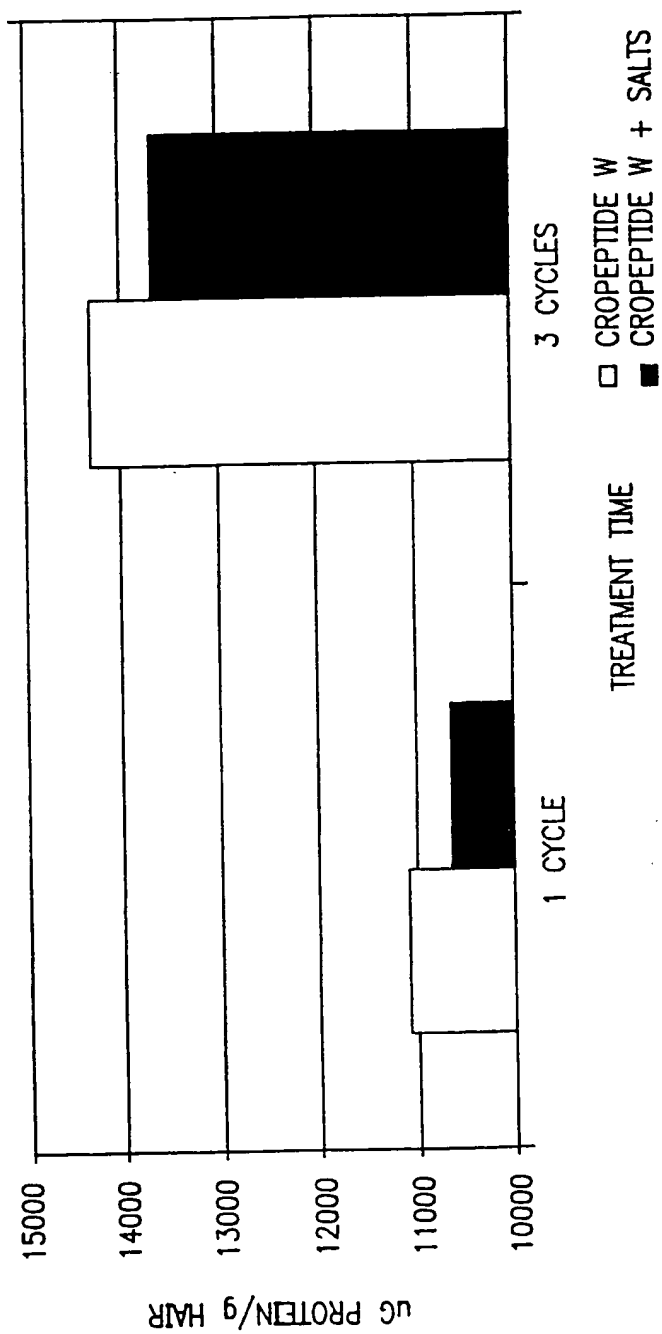
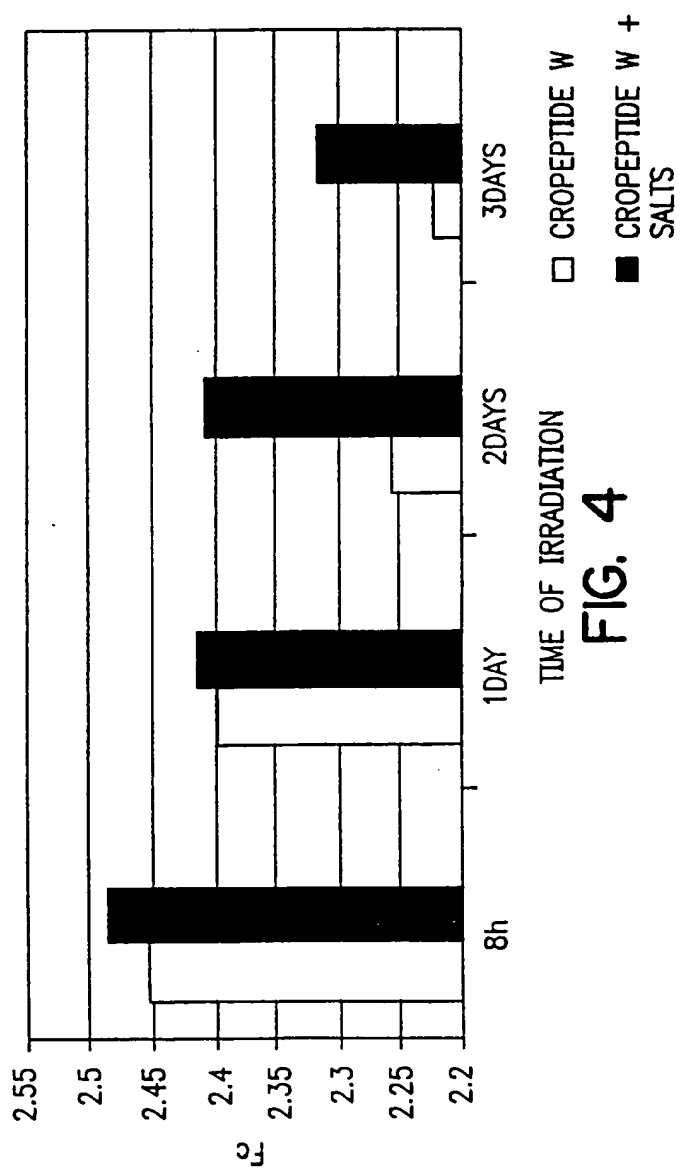


FIG. 3

7/11

**FIG. 4**

8 / 11

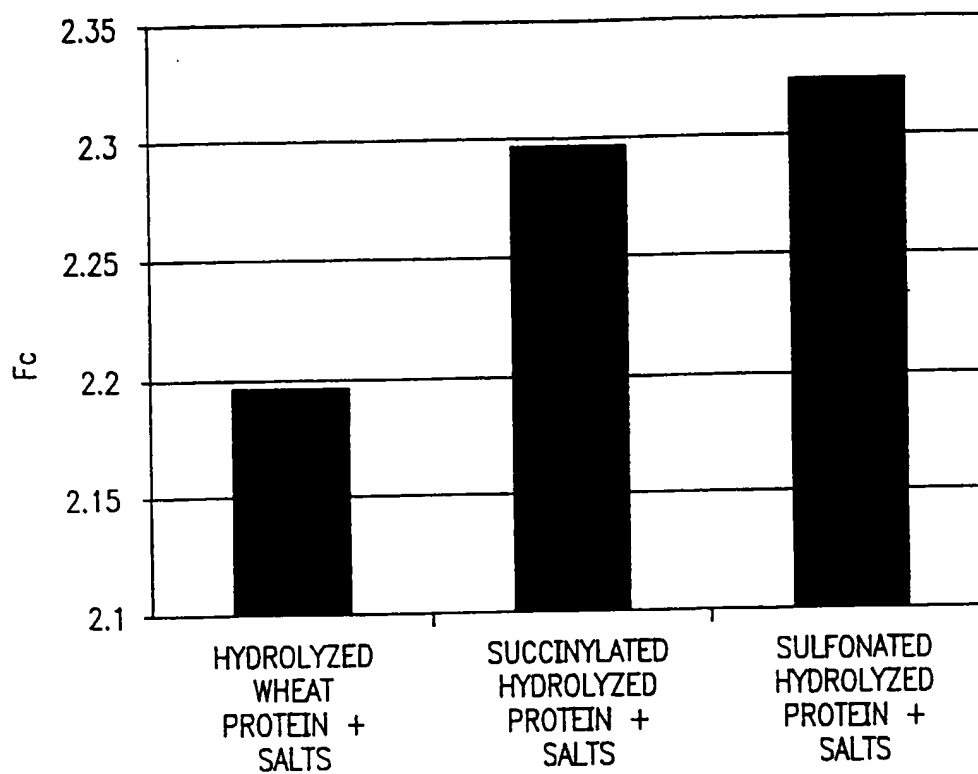
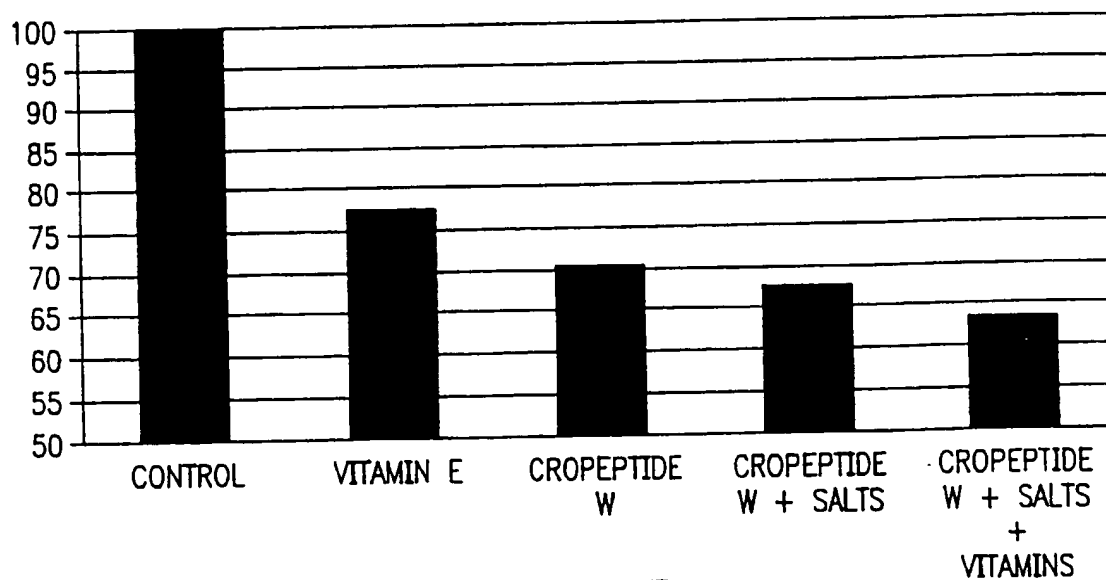


FIG. 5

TREATMENT
FIG. 6

9 / 11

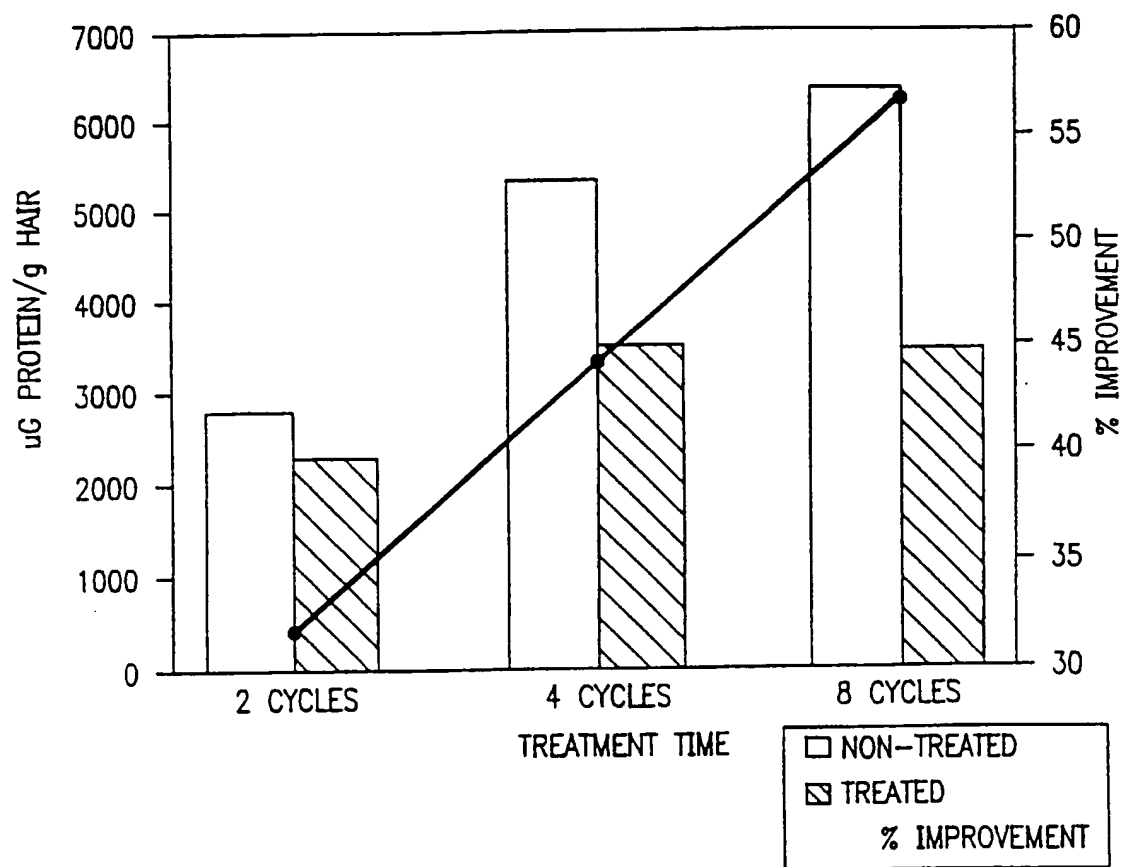
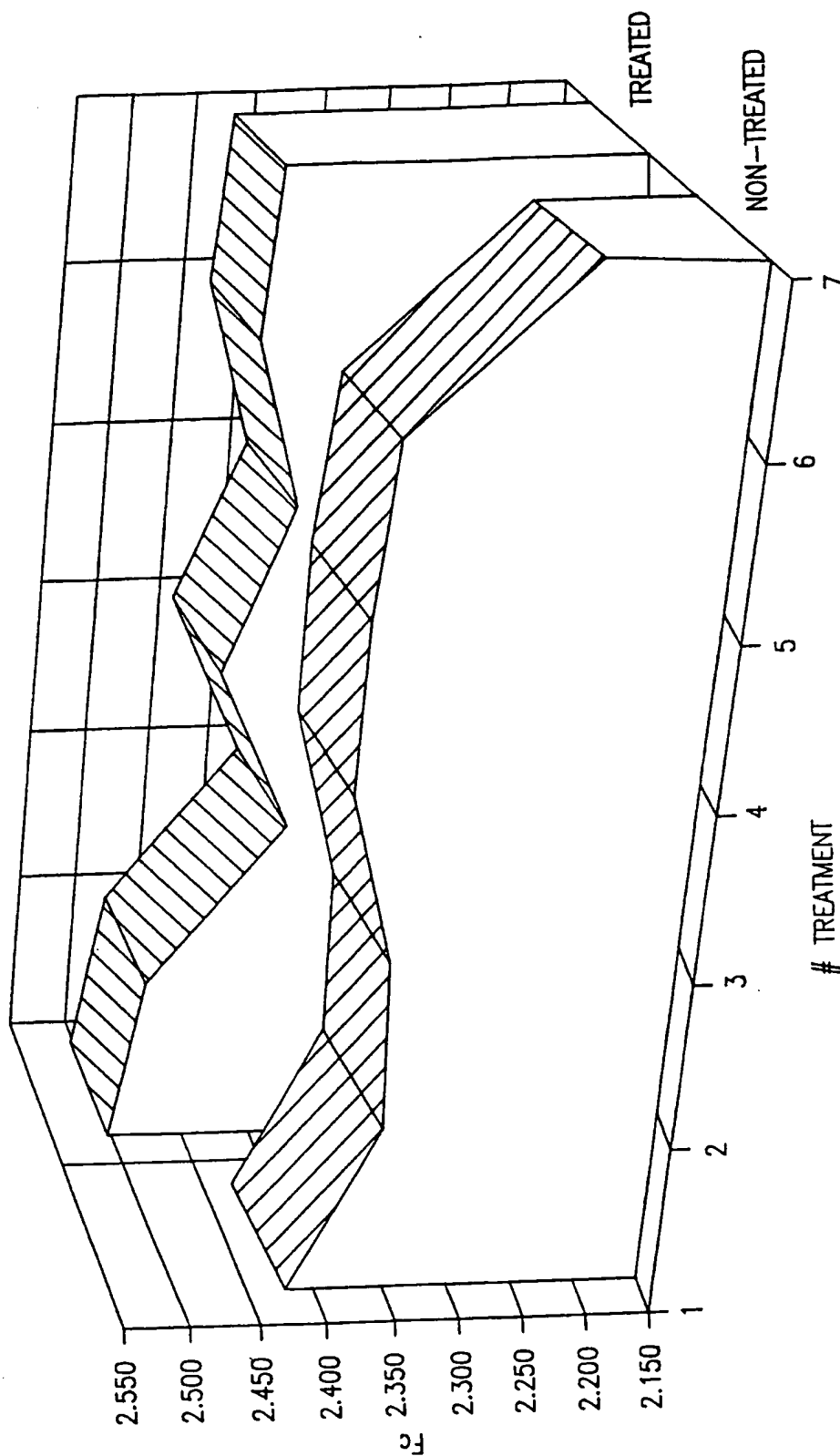


FIG. 7

10 / 11



TREATMENT
FIG. 8

11 / 11

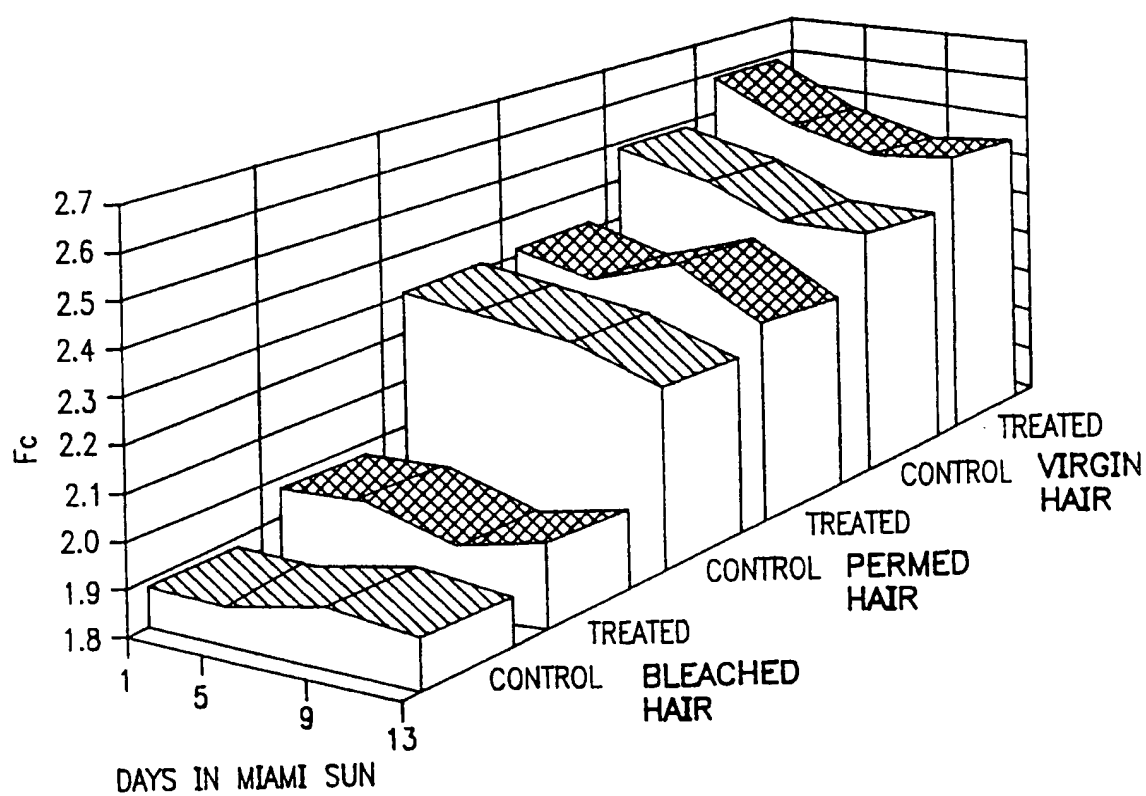


FIG. 9

INTERNATIONAL SEARCH REPORT

In tional Application No

PCT/US 97/19338

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 A61K7/06

According to International Patent Classification(IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 052 943 A (KAO SOAP CO., LTD) 2 June 1982 see the whole document ---	1-15
X	EP 0 057 837 A (KAO SOAP CO., LTD) 18 August 1982 see claim 1; examples 4,7,8 ---	1
A	EP 0 099 780 A (L'OREAL) 1 February 1984 see example 21 ---	1
A	DATABASE "CHEMICAL ABSTRACTS" (HOST: STN): Abs.124: 211 493, Columbus, OH, USA; & JP 07 330 560 A (SHISEIDO CO., LTD) 19 DECEMBER 1995 XP002056950 --- -/--	1

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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Date of the actual completion of the international search

25 February 1998

Date of mailing of the international search report

11/03/1998

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INTERNATIONAL SEARCH REPORT

In. tional Application No

PCT/US 97/19338

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	<p>DATABASE "CHEMICAL ABSTRACTS" (HOST: STN); Abs.124: 269 956, Colmbus, OH, USA; & JP 08 020 518 A (SHISEIDO CO., LTD) 23 JANUARY 1996 XP002056951</p> <p style="text-align: center;">-----</p>	1

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 97/19338

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